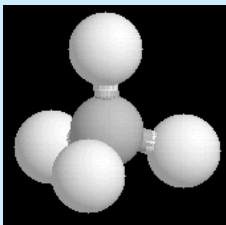
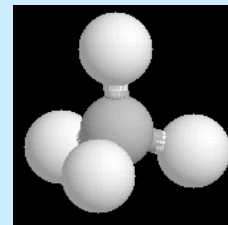


Infrared Laboratory Spectroscopy of CH₄ and CH₃D for Atmospheric Studies

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**The part of the research was performed at the Jet Propulsion Laboratory
(JPL), California Institute of Technology under contract with
the National Aeronautics and Space Administration
with many enthusiastic co-investigators**

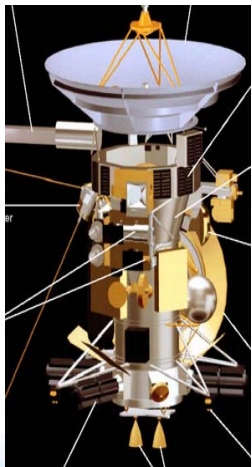
ASTRONOMICAL REMOTE SENSING



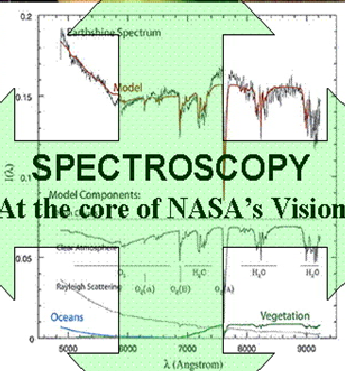
MARS
 $^{13}\text{C}/^{12}\text{C}$, H/D and $^{18}\text{O}/^{16}\text{O}$
 isotopic ratios serve as effective Martian biosignatures

VENUS
 Weak CO_2 features exploited to discern atmospheric dynamics

TPF
 Detect biosignatures, such as ozone or molecular oxygen, using visible and IR light



EARTH
 P, T and humidity retrievals depend on correct molecular line shape and continuum models

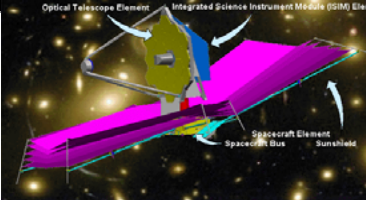


ASTROBIOLOGY



TITAN
 $^{15}\text{N}/^{14}\text{N}$ ratios indicate volcanic activity

OUTER PLANETS
 NIR CH_4 reveals atmospheric properties and structure



Basic transition line parameters

- **Line position** (or center frequency)
- **Line intensity @ 296 K**
- **Lower state energy** (for temperature dependence)
- Vibrational - rotational quantum assignment

Line shape parameters

- **Pressure-broadened widths** (usually Voigt)
& temperature dependence.
- **Line mixing**
- **Pressure-induced frequency shifts**
- **Self-broadened widths**
- Continua: collision-induced absorption (CIA)
(given as cross section files)

Line-by-line parameters should be **COMPLETE** and **ACCURATE (ENOUGH)**

ACCURACIES NEEDED FOR MANY APPLICATIONS

ν Positions: 0.00001 - 1.0 cm⁻¹ \

E'' Lower states energies (from assignments): $\approx 1/2\%$

S Line intensities: $\rightarrow 0.2$ to 10%

γ Pressure-broadening widths: $\rightarrow 0.2$ to 20 %

η Temperature dependence of widths: 10 to 40%

δ pressure-induced shifts: ??

METHODS TO OBTAIN SPECTROSCOPIC PARAMETERS

- **Calculations** based on **successful** theoretical modeling (good for positions and intensities, **but not line shapes**)
- **Predictions** based on **limited data and/or poorer theoretical modeling** (warning: extrapolations very poor!)
- **Empirical data** **retrieved line-by-line with some known assignments** (warning: no weak lines, larger uncertainties!)
- **Absorption cross sections** **from lab spectra, sometimes at different temperatures**
(for unresolved heavy species and continua)

Near-IR CH₄ (and CH₃D): Positions and Intensities:

Difficult to model because energy levels perturb each other.

Triacontad: intractable:
empirical linelist with 1% assigned.

Icosad: almost intractable; one strong
band studied, empirical list

Tetradecad: mostly unassigned; no
public prediction; empirical linelist

Octad: new predictions coming for main
isotope only.

CH₃D overtones/combination →

Pentad: 3 isotopes (fundamentals and
overtones) modeled;

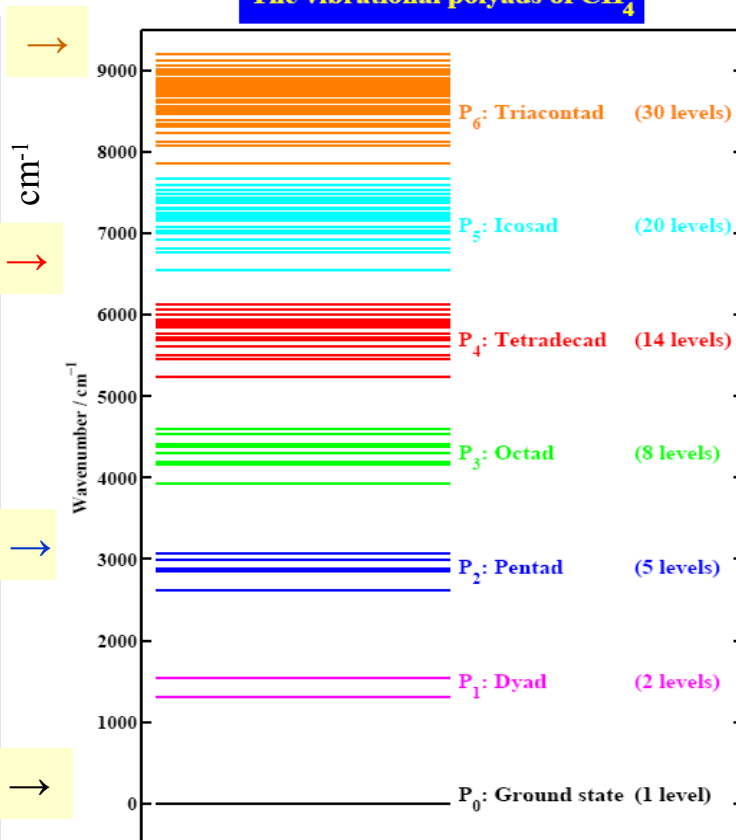
Hot bands intensities are estimated.

Dyad: 3 isotopes good. Hot bands
intensities modeled to 8%.

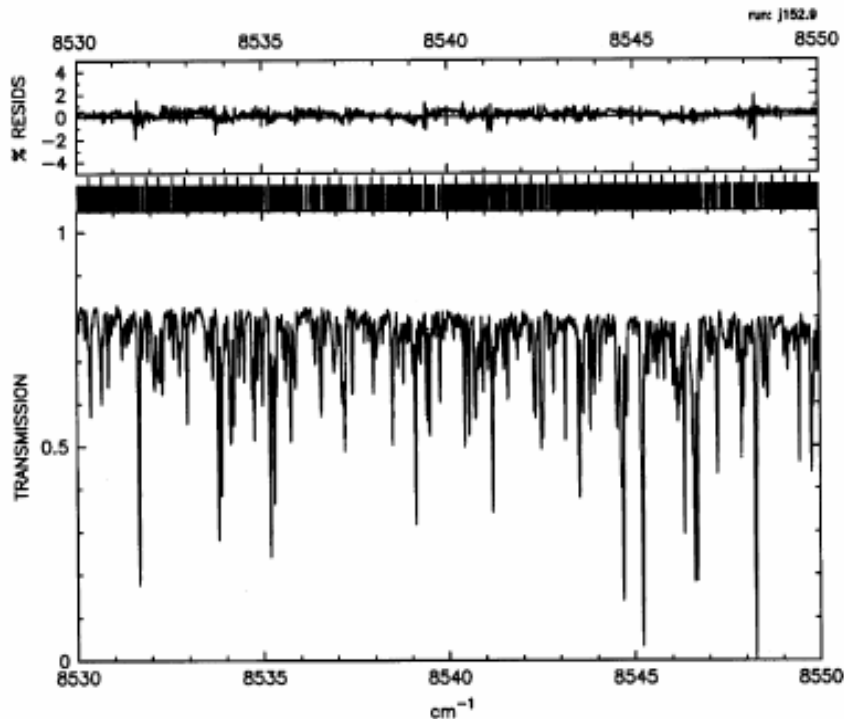
GS: Intensities low by 15%.

CH₃D prediction is old.

The vibrational polyads of CH₄



Triacontad



Room temperature methane lab spectrum at 0.022 cm^{-1} resolution
Retrieved 537 positions and intensities Line-by-line

Bottom panel is observed spectrum overlaid with synthetic; top is difference plot.

**No known assignments: can not compute spectra at other temperatures
 or attach broadening coefficients as a function of quantum numbers**

Highly accurate potential-energy and dipole moment surfaces for vibrational state calculations of methane

- ◆ **Vibrational configuration-interaction calculations of energy levels and the absorption intensities up to 9000 cm^{-1}**
- ◆ **full-dimensional ab initio potential-energy surface (PES) and dipole moment surface for methane**
- ◆ **Predicted energy levels within 10 cm^{-1} of observed**

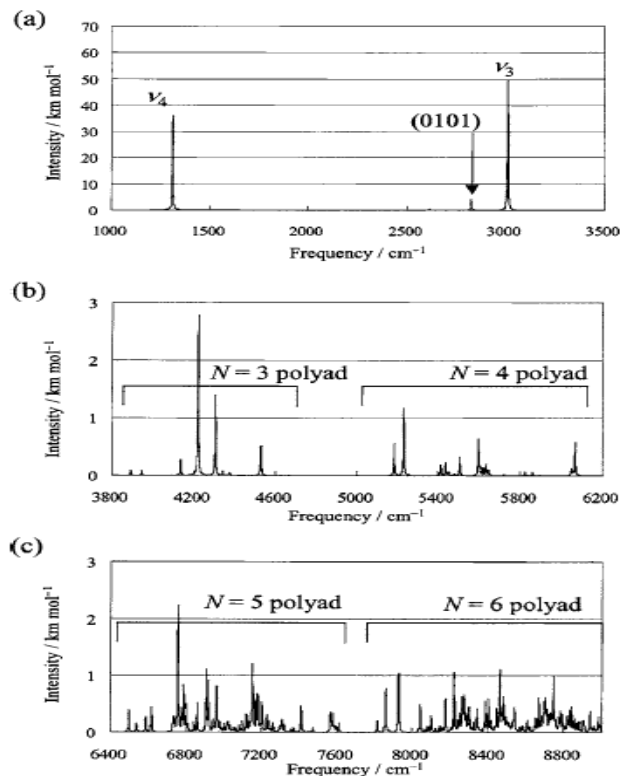


FIG. 2. Theoretical vibrational spectra obtained from VCI calculations using the MSI(4th) potential-energy functions and dipole moment surfaces. Panels (b) and (c) are scaled up for clarity.

Identification of hot band transitions of CH₄ near 3000 cm⁻¹

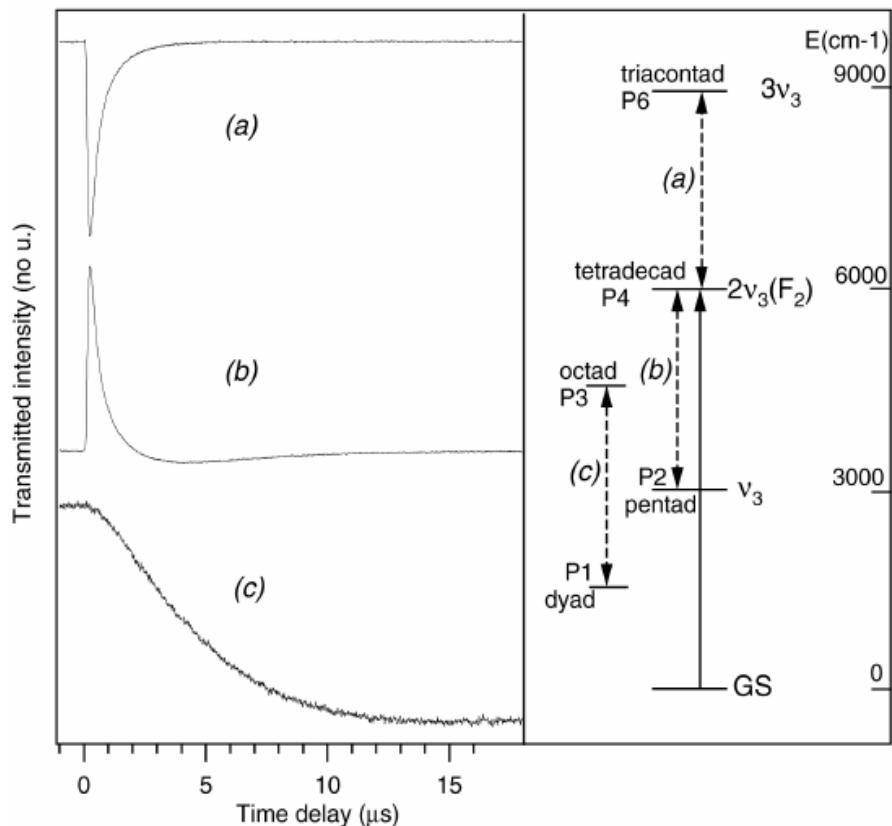
C. Boursier*, J. Menard, A. Marquette, F. Menard-Bourcin

Figure 1

Double resonance
experiment

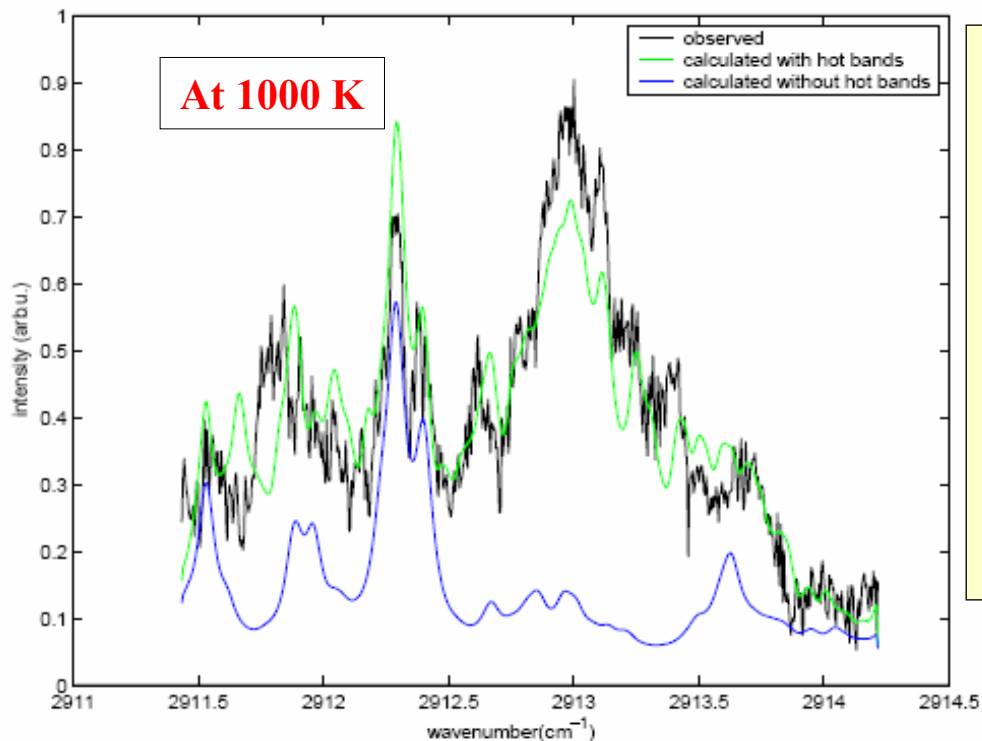
Assign higher
states via
hot band
transitions.

Accepted J. Mol.
Spectrosc.



◆ Methane Hot bands Are Important

The methane Raman spectrum from 1200 to 5500 cm^{-1} :
A first step toward temperature diagnostic using
methane as a probe molecule in combustion systems



Jourdanneau

et al.

J. Mol. Spectrosc.

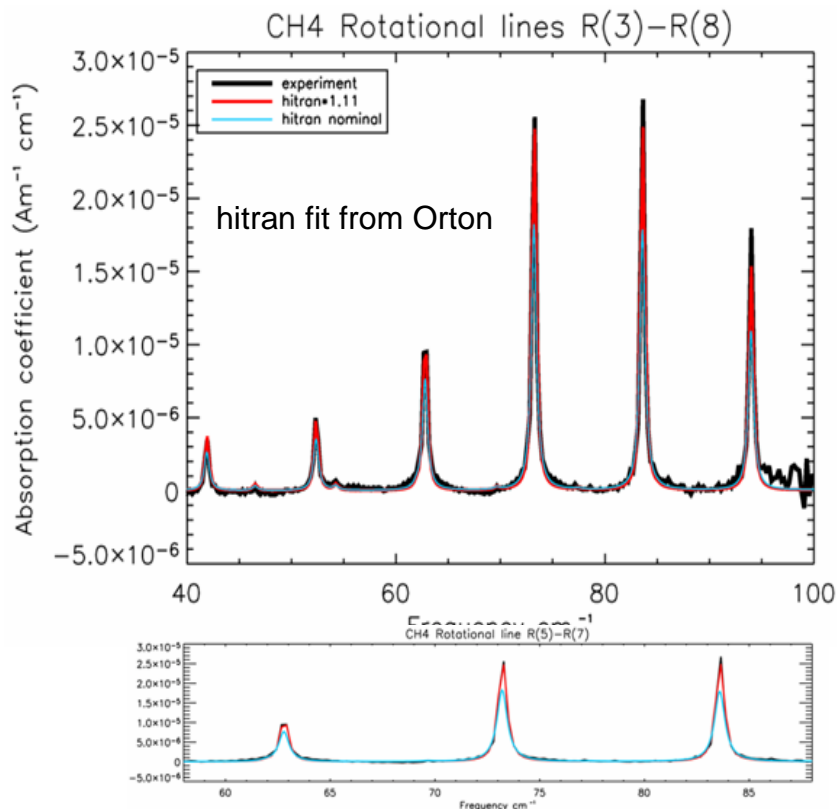
Vol. 233

Pgs 219-231

(2005)

Fig. 10. Simulation without, and with hot bands in the region 2911–2914 cm^{-1} . The band centered at 2913 cm^{-1} is $\nu_1 + \nu_4 - \nu_4$.

Far-IR CH₄ Intensities for ground state transitions in HITRAN and GEISA low by 15% ± ?



Lab Spectra of Far-IR CH₄ (Wishnow)

HITRAN intensities for Far IR set by one “indirect method”, (calc.) [Hilico et al., J Mol Spec, 122, 381 (1987)] with claim of accuracy of $\pm 30\%$.

Cassam-Chenai, [JQSRT, 82,251(2003)] predicts ab initio Q branch based on Stark measurements [Ozier et al. Phys Rev Lett, 27, 1329, (1971)].

The intensities are 15% higher than HITRAN values.

Lab data (left) confirms a higher value for R branch manifolds.

Models for Collision-Induced Continua

- ◆ <http://www.astro.ku.dk/~aborysow/programs/>

Borysow & Frommhold calculate collision-induced spectra at different temperatures and then form model spectra of cross sections.

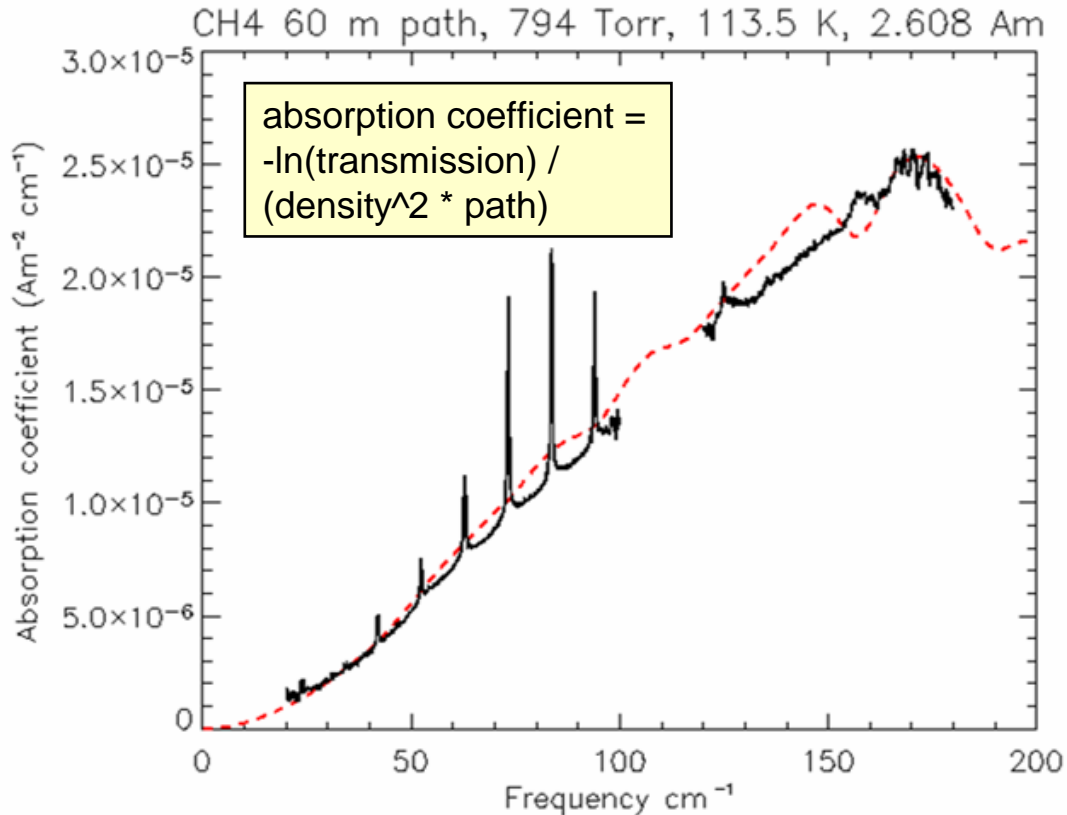
- ◆ Very useful models and software available for generating synthetic spectra

$\text{H}_2\text{-H}_2$, $\text{H}_2\text{-He}$, $\text{H}_2\text{-CH}_4$, $\text{H}_2\text{-Ar}$, $\text{N}_2\text{-N}_2$,

$\text{CH}_4\text{-CH}_4$, $\text{N}_2\text{-CH}_4$, $\text{CH}_4\text{-Ar}$, $\text{CO}_2\text{-CO}_2$

New work in progress for $\text{H}_2\text{-H}_2$ by Gustafsson and Frommhold

Low temperature spectrum of methane



First observation of
R(3)-R(7) lines

Wishnow, Leung, Gush,
Rev. Sci. Inst., 70, 23
(1999) at 0.24 and 0.06
 cm^{-1} resolution

Ground State
transitions with calc.
Collision-Induced
absorption shows
need to improve the
CIA parameters

Dashed line: CH₄ Collision-Induced Absorption
(CIA) from Borysow.

Air- and Self-broadened Widths of Methane 5 Bands of the OCTAD: 4100 to 4635 cm^{-1}

Widths vary as a function of quanta and band symmetry

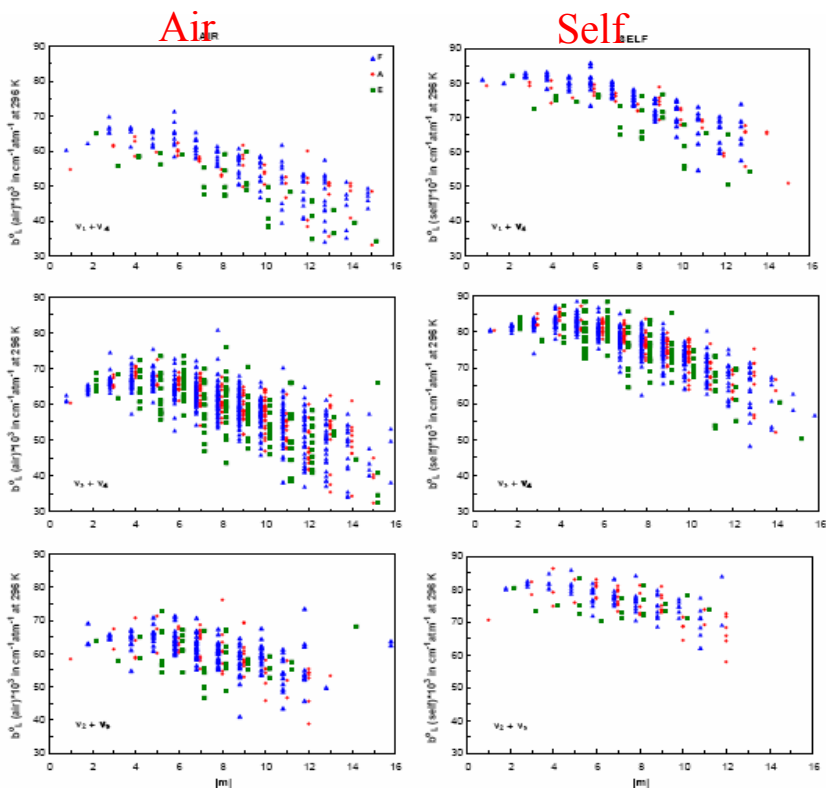


Fig. 3. Measured air-broadening and self-broadening coefficients, b_{ν}^{\pm} (in $\text{cm}^{-1} \text{atm}^{-1}$ at 296 K), in the three strongest bands ($\nu_1 + \nu_4$, $\nu_3 + \nu_4$, and $\nu_2 + \nu_3$) of the octad polyad plotted as a function of $|m|$, where $|m|$ is the lower state J for P and Q transitions and the upper state J for R branch lines. The error bars are not plotted to allow a better observation of the trends in broadening coefficients as a function of $|m|$.

$\nu_1 + \nu_4$ at 4220 cm^{-1} :

3-fold degenerate (F2).

These widths are within 4% of ν_3 values (at 3020 cm^{-1}) and other bands with a 3-fold vibrational symmetry (F2).

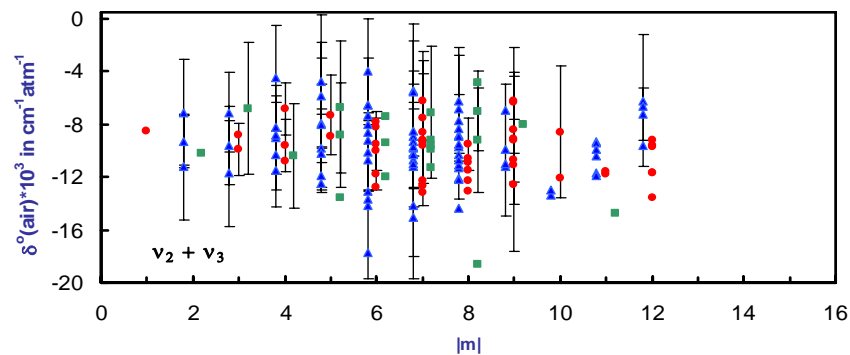
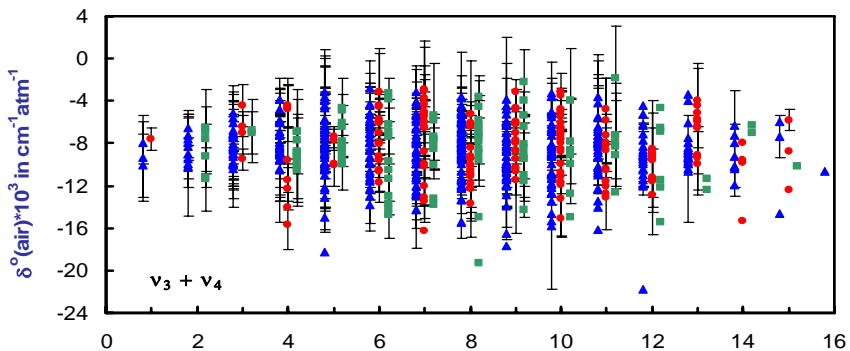
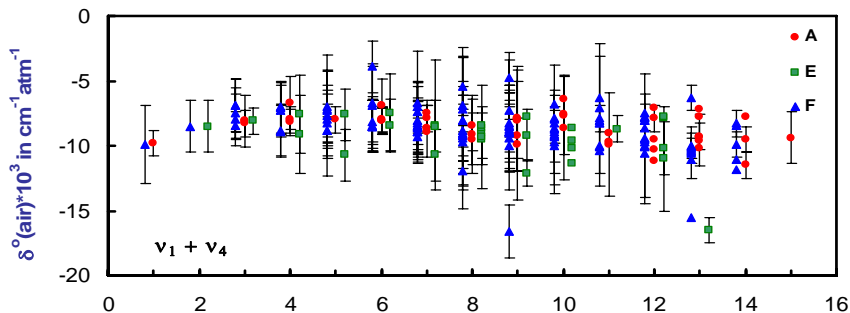
$\nu_3 + \nu_4$ at 4310 cm^{-1} :

9-fold degenerate band: variation of widths at each J is much greater.

$\nu_2 + \nu_3$ at 4530 cm^{-1} :

6-fold degenerate band: some variation of widths at each J .

Predoi-Cross et al. J. Mol. Spect.



**Air-broadened
pressure shifts: δ**

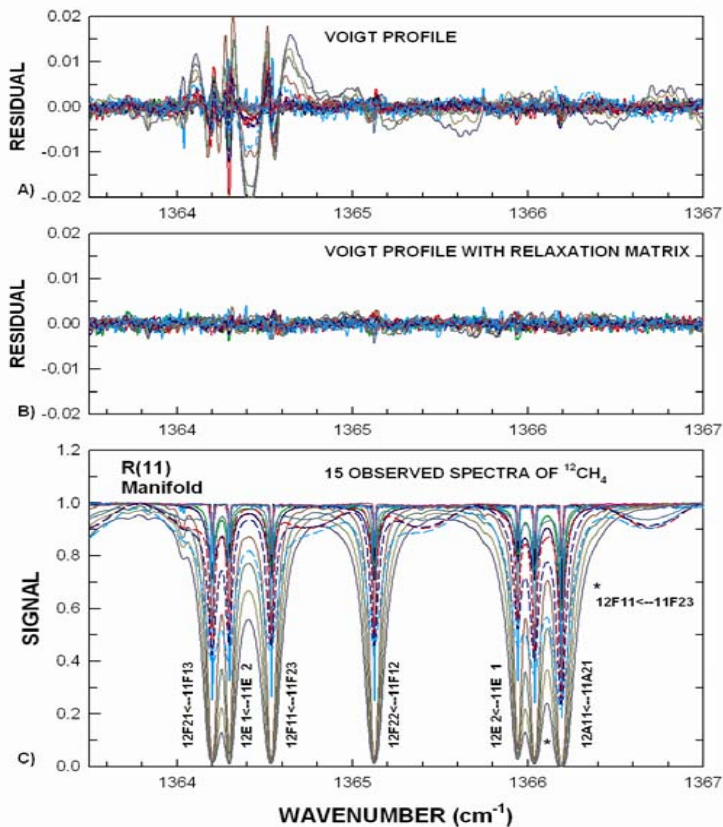
**NO obvious patterns
vs quantum numbers**

**Pressure shifts can be
estimated with a simple
linear expression:**

**$\delta \sim -1.97(24) \times 10^{-6} \times v$
where v is the vibrational
band center.**

**Is the scatter evidence
of line mixing?**

Studies to Understand Line Mixing in CH₄



Multispectrum fitting of line mixing in self- and air-broadened ν_4 methane (M. Devi)

Line Mixing Rules

A1 ↔ A2 **but not** A2 ↔ A2

F1 ↔ F2 **but not** F2 ↔ F2

E ↔ E

Sum of mixing coefficients = 0.

First noticed in Raman spectra
 Stimulated Raman $2\nu_2$ Q-branch,
 (Millet et al. 1991)

Most absorption studies at 3000 cm⁻¹

1992-2003: Pine - **measurements**

1999-2001: Pieroni et al. - **theory**

2001-2002: Grigoriev – **high pressure**

2006: Tran, Hartmann et al.

(Two papers in press JQSRT)

Models at very high pressure
 in both ν_4 and ν_3 **with software**
and database for air, N₂, H₂, Ar & He

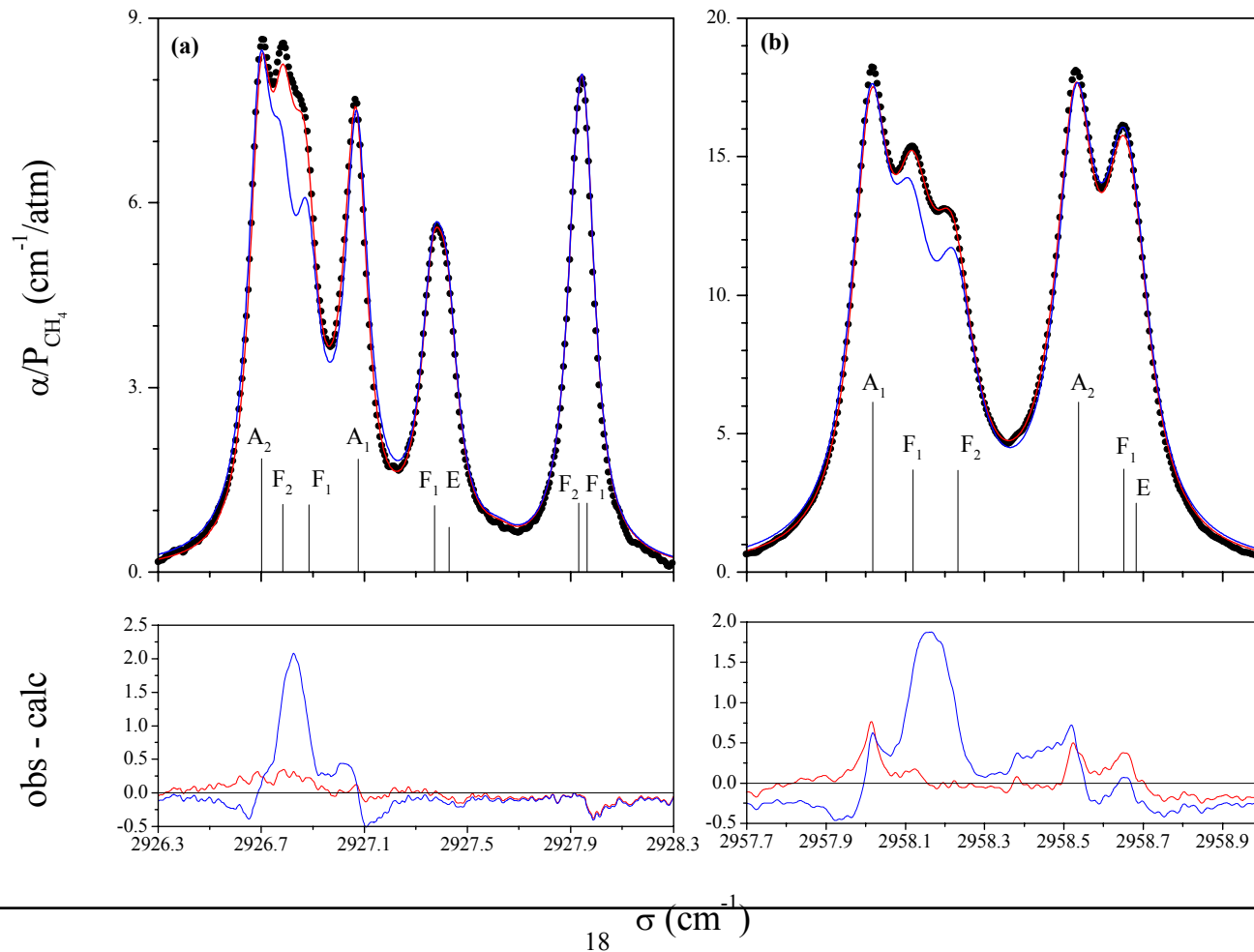
Collisional effects in CH₄ spectra, IR, Raman, and Atmospheres

Jean-Michel Hartmann,

Laboratoire de PhotoPhysique Moléculaire

- Individual IR line shapes
- Shapes of IR manifolds
- Shapes of IR branches and bands
- The IR wings
- The case of Raman spectra
- Collision induced absorption
- Effects on atmospheric transmission
- Effects on Jovian Emission

ν_3 band, P(9) and P(6) manifold, CH₄-N₂ at 300 K, 1 atm



The “Enneadecad” of CH₃D: 3300 to 4900 cm⁻¹

A. Nikitin, Laboratory of Theoretical Spectroscopy, Tomsk RUSSIA

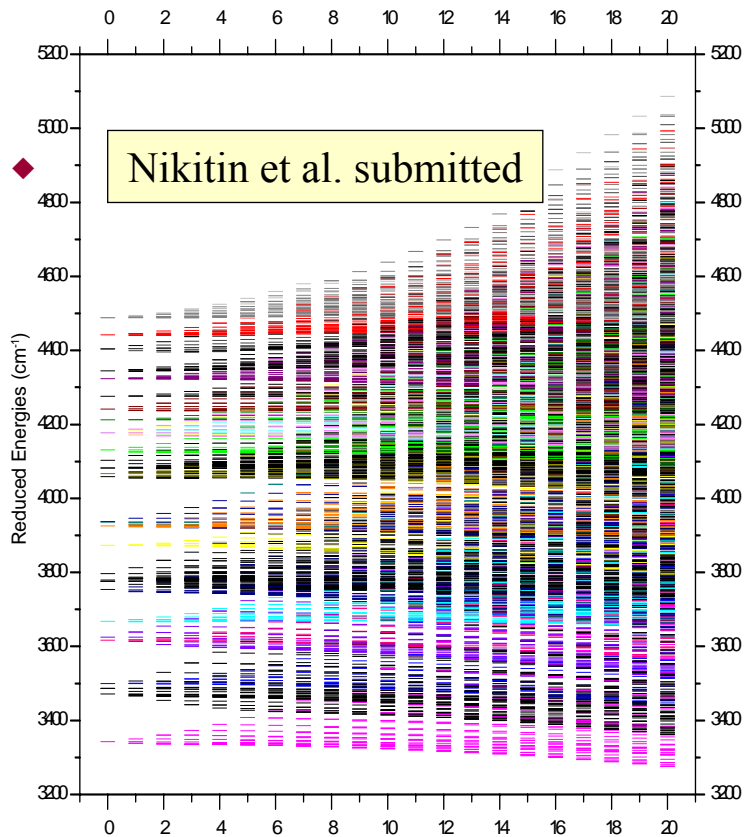
L.R. Brown, Jet Propulsion Laboratory, Pasadena USA

J.P. Champion, Laboratoire de Physique, Dijon France

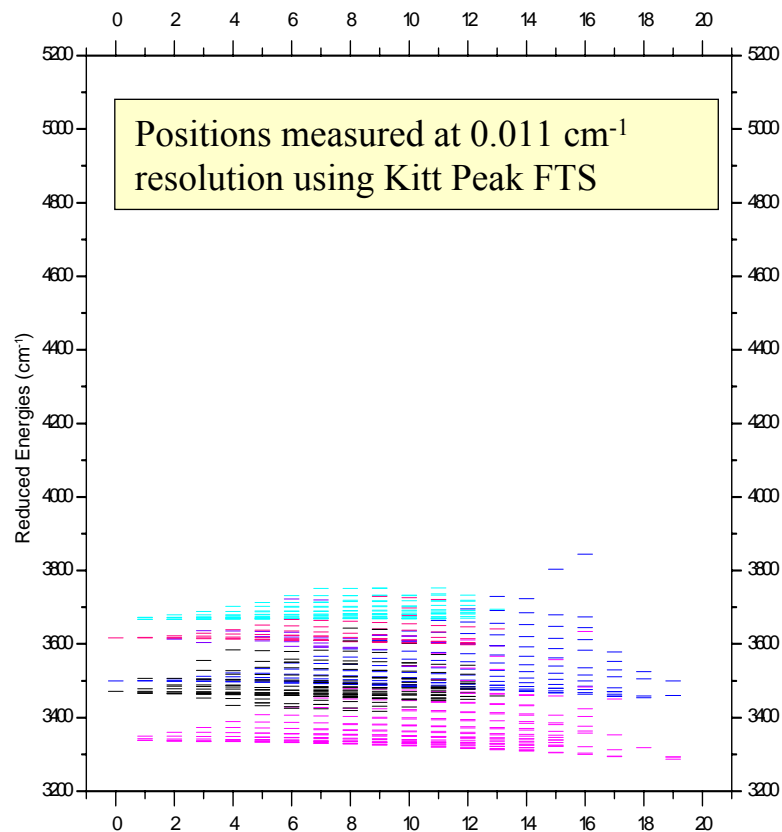
- **Enneadecad = 19 vibrational states (37 vibrational sublevels)**
- **Global analysis of 8 sublevels at 2.9 μm & hot bands at 3.3 μm**
- **Measured Intensities between 3300 – 3700 cm⁻¹**
- **441 effective Hamiltonian parameters fixed to values known from prior analyses of the Ground State, Triad and Nonad of CH₃D**
- **144 extra effective Hamiltonian parameters fitted using newly assigned “Enneadecad” and “Enneadecad – Triad” hot transitions**
- **0.0008 cm⁻¹ Fit of positions (6th order) : 2702 cold & hot band transitions**
- **4.5 % Fit of intensities (3rd order) : more than 850 measurements**

The Enneadecad of CH₃D Energy Levels

Calculated

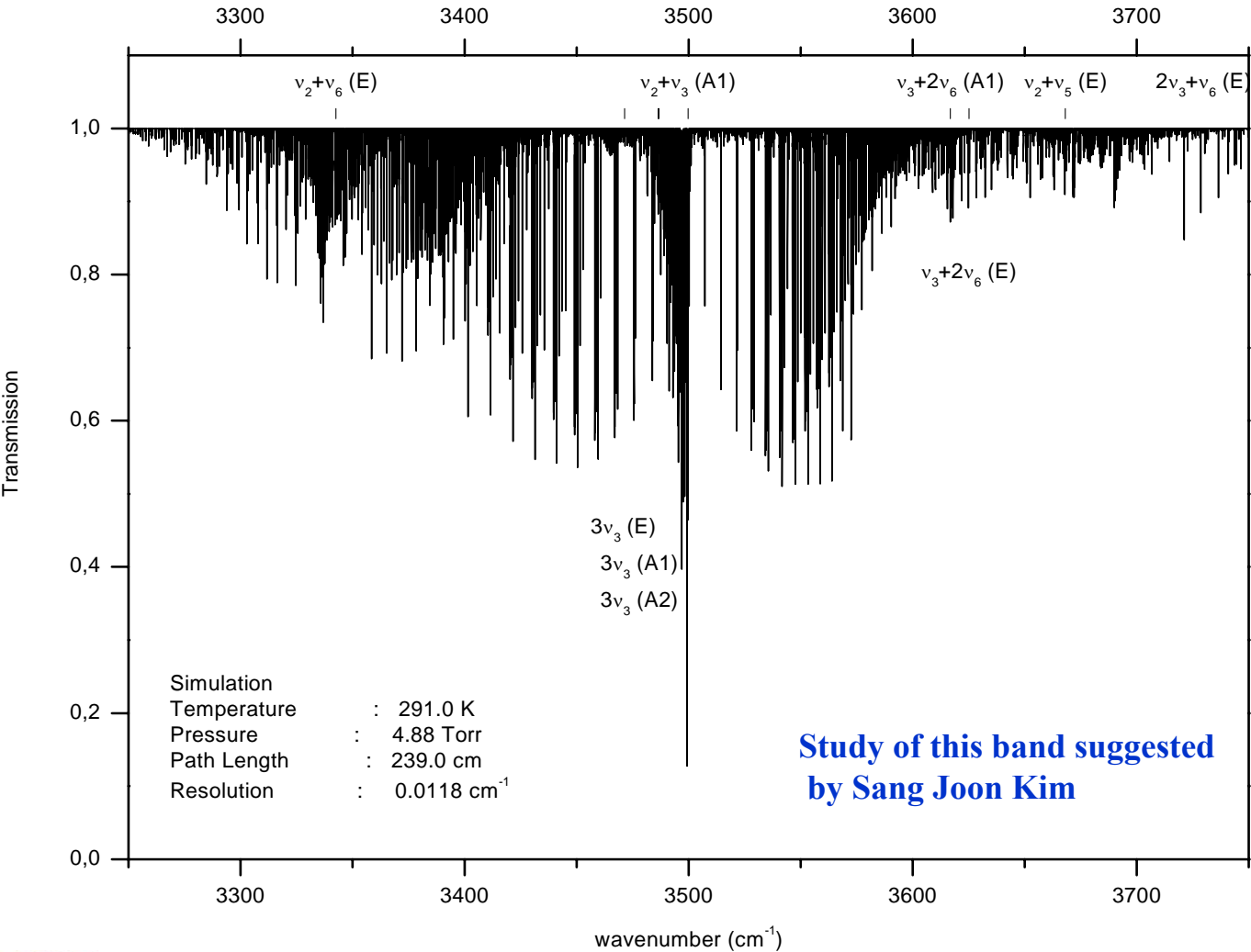


Observed



Reduced Upper State Energies (cm⁻¹) vs Quantum number J

Spectrum Overview (Simulation)



Recent Research: pressure broadening of CH₃D

Extensive measurements of four lowest fundamentals

Several broadeners: Self, O₂, N₂, air, H₂

Empirical expressions as a function of J,K

Theoretical calculations of widths beginning to match data

- ◆ **Measurements and [theoretical calculations](#) of N₂-broadening and N₂-shift coefficients in the ν_2 band of CH₃D**, Predoi-Cross et al. J. Mol. Spectrosc. 235 (1): 35-53, 2006
- ◆ **Measurements and [theoretical calculations](#) of self-broadening and self-shift coefficients in the ν_2 band of CH₃D**, Predoi-Cross et al. J. Mol. Spectrosc. 234 (1): 53-74, 2005
- ◆ **H₂-broadening coefficients in the ν_3 band of CH₃D at low temperatures**
Lerot et al. J. Mol. Spectrosc. 219 (2): 329-334, 2003
- ◆ **Diode-laser measurements and calculations of H₂-broadening coefficients in the ν_3 band of CH₃D**. Lerot et al. J. Mol. Spectrosc. 217 (1): 79-86, 2003
- ◆ **Self- and N₂-broadening, pressure induced shift and line mixing in the ν_5 band of CH₃D using a multispectrum fitting techniques**. Devi et al. J. Quant. Spectrosc. & Rad. Transfer 74 (1): 1-41, 2002 plus papers for air-broadening. [Modeled with empirical expressions](#).

Room-temperature broadening and pressure-shift coefficients in the ν_2 band of $\text{CH}_3\text{D}-\text{O}_2$: Measurements and semi-classical calculations

Adriana Predoi-Cross ^{a,*}, Kyle Hambrook ^a, Shannon Brawley-Tremblay ^a, Jean-Pierre Bouanich ^b, V. Malathy Devi ^c, Mary Ann H. Smith ^d

A. Predoi-Cross et al. / Journal of Molecular Spectroscopy 236 (2006) 75–90

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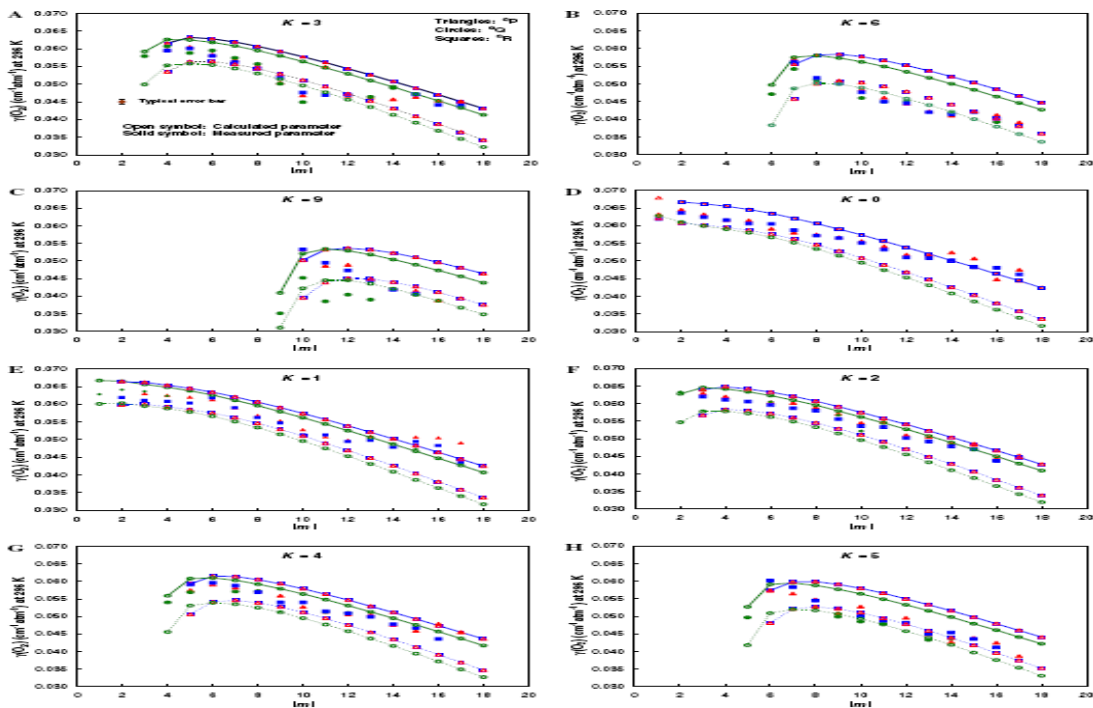


Fig. 5. Variation of measured, empirically derived, and theoretically calculated O_2 -broadening coefficients with $[\text{O}_2]$ for (A) $K = 3$; (B) $K = 6$; (C) $K = 9$; (D) $K = 0$; (E) $K = 1$; (F) $K = 2$; (G) $K = 4$; and (H) $K = 5$. For a given $[\text{O}_2]$, the theoretical broadening coefficients for the V^P and V^R lines are very nearly the same. The broadening coefficients connected with smooth line are obtained using calculation 1 and those connected with dotted line are obtained using calculation 2.

Spectroscopy Needed for CH₄ and CH₃D

- ◆ **Far-IR:** Improved accuracy for line-by-line intensities
- ◆ **Far-IR:** Better models for CIA with different broadeners
- ◆ **Mid-IR:** Analysis of weaker hot bands (Octad-Dyad,...)
- ◆ **Near -IR:** Extensive analyses for all isotopes
- ◆ **IR:** Models for pressure broadening coefficients
- ◆ **Near-IR:** Extensive broadening measurements
- ◆ **IR:** Line mixing measurements and models